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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.002 Å R factor = 0.039 wR factor = 0.108 Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Half of the title compound, $C_{38}H_{44}O_2$, makes up the asymmetric unit; the whole molecule possesses inversion symmetry. The calix[4]arene conformation is similar to the unsubstituted calix[4]arene. Only one intermolecular interaction is noted.

5,17-Di-tert-butyl-25,27-dihydroxycalix[4]arene

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Comment

The title compound was prepared as part of our investigations into the synthesis of new calix[4]arene-based crown ether extractants (Sachleben et al., 1999). Only half of the molecule makes up the asymmetric unit; the other half is generated by crystallographic inversion the centre [symmetry code: 1 - x, 1 - y, 1 - z, denoted (i) throughout this paper]. The conformation of a calix[4]arene is generally defined in terms of the relative orientation of the arene rings with respect to the best plane defined by the four calix methylene-C atoms (C17, C18, C17ⁱ, and C18ⁱ). This methylene carbon plane is referred to as the reference plane (Perrin & Oehler, 1991). The planes of the two methoxy-substituted arene rings (A1 and $A1^{i}$) are nearly perpendicular to the reference plane $[84.5 (1)^{\circ}]$, while the unsubstituted arene rings (A2 and A2ⁱ) appear to be roughly parallel to it $[173.9 (1)^{\circ}]$. This calix configuration closely resembles that of the unsubstituted parent molecule, calix[4]arene (McMurry & Phelan, 1991) and that of tetra-tert-butyl-calix[4]arene (Grynszpan et al., 1991). Strangely, the more closely related tetra-tert-butyl-diethoxycalix[4]arene adopts a 1,3-alternate calix conformation (Ting et al., 1995).



Examination of the extended structure reveals only one intermolecular interaction worthy of mention, an edge-face arene interaction between A2 and A1. The H16 \cdots A1_{centroid} distance is 2.64 Å and the C16-H16 \cdots A1_{centroid} angle is 160°.

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Figure 1

The molecular structure of the title compound showing 50% displacement ellipsoids. Symmetry code: (i) 1 - x, 1 - y, 1 - z.

Experimental

The title compound was prepared by alkylation of bis-9,12-*tert*-butyl-25,27-dihydrocalix[4]arene using methyl iodide and potassium carbonate in refluxing acetonitrile. Crystals suitable for X-ray diffraction grew from fractions obtained following chromatography (SiO₂ ether/hexanes eluent).

Crystal data

C ₃₈ H ₄₄ O ₂	$D_x = 1.187 \text{ Mg m}^{-3}$
$M_r = 532.7$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 25
a = 8.6623 (8) Å	reflections
b = 18.1338 (13) Å	$\theta = 10.1 - 14.3^{\circ}$
c = 9.8554 (9) Å	$\mu = 0.07 \text{ mm}^{-1}$
$\beta = 105.68 \ (8)^{\circ}$	$T = 100 { m K}$
$V = 1490.5 (2) \text{ Å}^3$	Fragment, colourless
Z = 2	$0.70 \times 0.41 \times 0.30 \text{ mm}$
Data collection	
Nonius CAD-4 diffractometer	$\theta_{\rm max} = 25.0^{\circ}$
ω scans	$h = 0 \rightarrow 10$
Absorption correction: none	$k = -10 \rightarrow 21$
3130 measured reflections	$l = -11 \rightarrow 11$
2618 independent reflections	3 standard reflections
2090 reflections with $I > 2\sigma(I)$	frequency: 120 min
$R_{\rm int} = 0.014$	intensity decay: 5%

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0547P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	+ 0.4494P]
$wR(F^2) = 0.108$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} = 0.001$
2618 reflections	$\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$
185 parameters	$\Delta \rho_{\rm min} = -0.22 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Sel	lected	geometric	parameters	(A, °	').	
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0-C1	1.3891 (18)	O-C19	1.432 (2)
C1-O-C19 O-C1-C2	115.01 (14) 119.92 (15)	O-C1-C6	119.19 (15)

A 1.1 mm collimator was used. All H atoms were placed in calculated positions, refined using a riding model, and given an isotropic displacement parameter equal to 1.2 (CH, CH₂) or 1.5 (CH₃) times the equivalent isotropic displacement parameter of the atom to which they were attached. The C-H distances used depended on the type of C atom: $C_{methylene}$ -H = 0.99, C_{methyl} -H = 0.98, and C_{arene} -H = 0.95 Å. Methyl-H atoms were allowed to rotate about the adjacent C-C bond.

Data collection: *CAD*-4-*PC* (Enraf–Nonius, 1993); cell refinement: *CAD*-4-*PC*; data reduction: *XCAD*-4 (Harms, 1995); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *PLATON* (Spek, 2001).

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addenda and errata

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5,17-Di-*tert*-butyl-25,27-dihydroxycalix[4]arene. Erratum

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In the paper by Bryan, Sachleben & Urvoas [*Acta Cryst.* E**57**, o781–o782], the chemical name in the title is incorrect. The correct title is 5,17-di-*tert*-butyl-25,27-dimethoxycalix[4]arene.

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